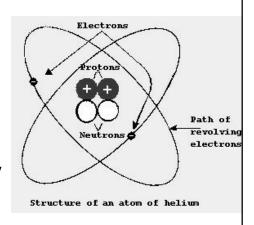
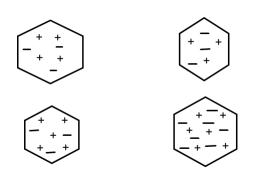


## To understand Static you must understand The Atom

- All matter is made of atoms that are combined together into molecules
- The atom is composed of protons, neutrons and electrons
  - Protons have a positive charge, -can't move
  - Electrons have a negative charge -lost or gained easily
  - > Neutrons are neutral



#### **Determine the Charge on the Following Objects**



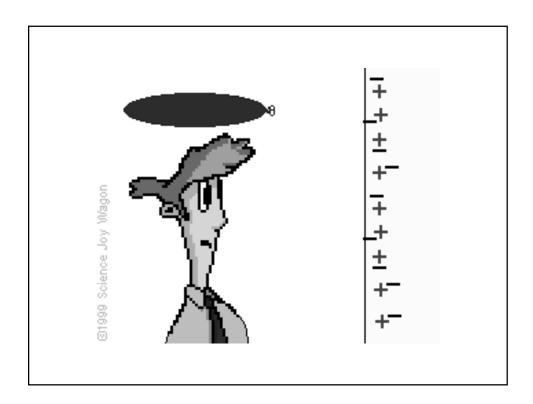
How would you make each of the above neutral?

### **Static Electricity**

Static electricity is created by applying a friction force between two objects

The friction force transfers electrons from one object to the other

Static Electricity is the study of charges at rest.



#### **Static Electricity**

If something gains electrons It has an excess of electrons, it is negatively charged

If something loses electrons it has a deficiency of electrons It is positively charge.

Neutral objects have equal numbers of protons and electrons

#### **Predicting Charge**

- You can predict what charge an object gets by using the electrostatic series
- O Q 1 If you rub silk with fur what charge will each get?
- Q 2: If silk is rubbed on plastic what charge will each object get?

#### **Electrostatic Series**

glass
your hair
nylon
wool
fur
silk
paper
cotton
ebonite
hard rubber
polyester
polyvinylchloride plastic

Greater attraction for electrons

#### **Laws of Electrostatics**

- Like charges repel.

  ➤ Positive repels positive

  ➤ Negative repels negative

#### Opposites attract.

▶ Positives attract negative

# Charged objects attract neutral objects. Negative attracts neutral Positive attracts neutral

